

Electrolyser Technologies: Challenges and Opportunities

Theme: Hydrogen Production – Electrolysis & Biopathways



05th - 07th July 2023, Vigyan Bhawan, New Delhi

Prof. Kaliaperumal Selvaraj
National Chemical Laboratory



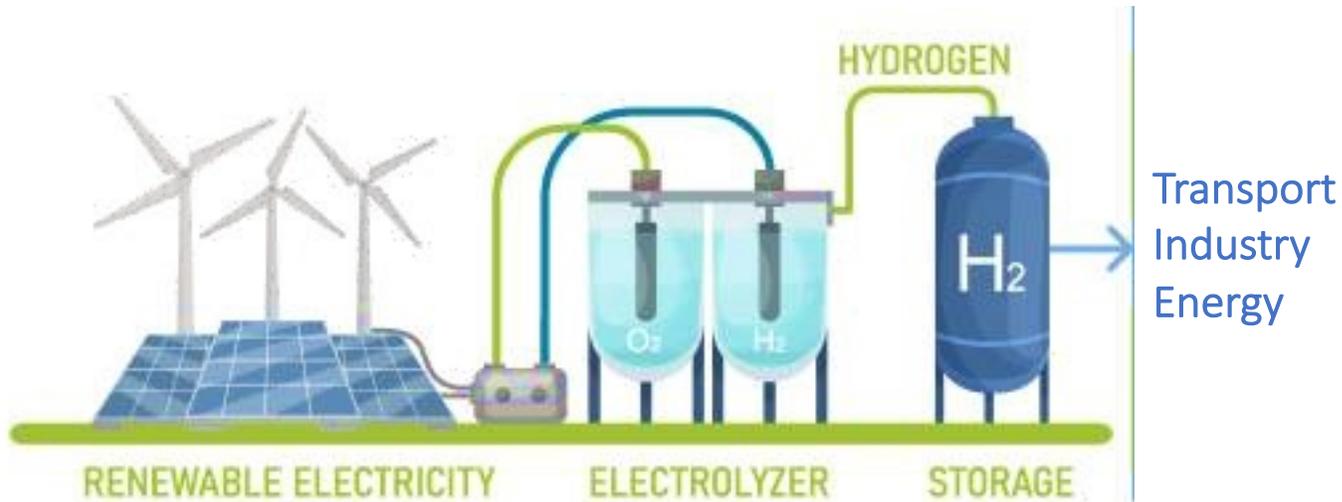
Council of Scientific and Industrial Research
Ministry of Science and Technology



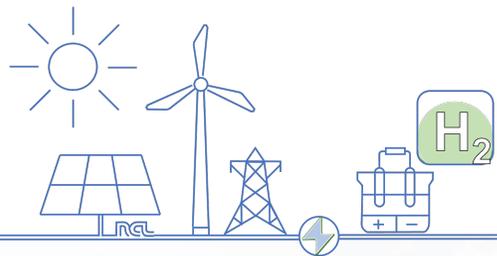
Prelude: Outline & Scenario



Hydrogen Economy Energy Transition Affordable Green Hydrogen World, India & CSIR

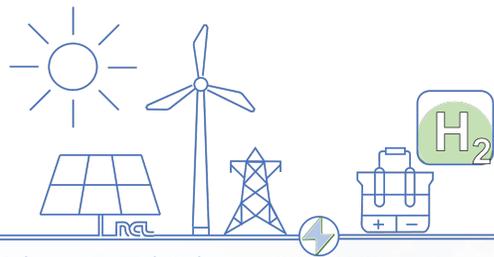
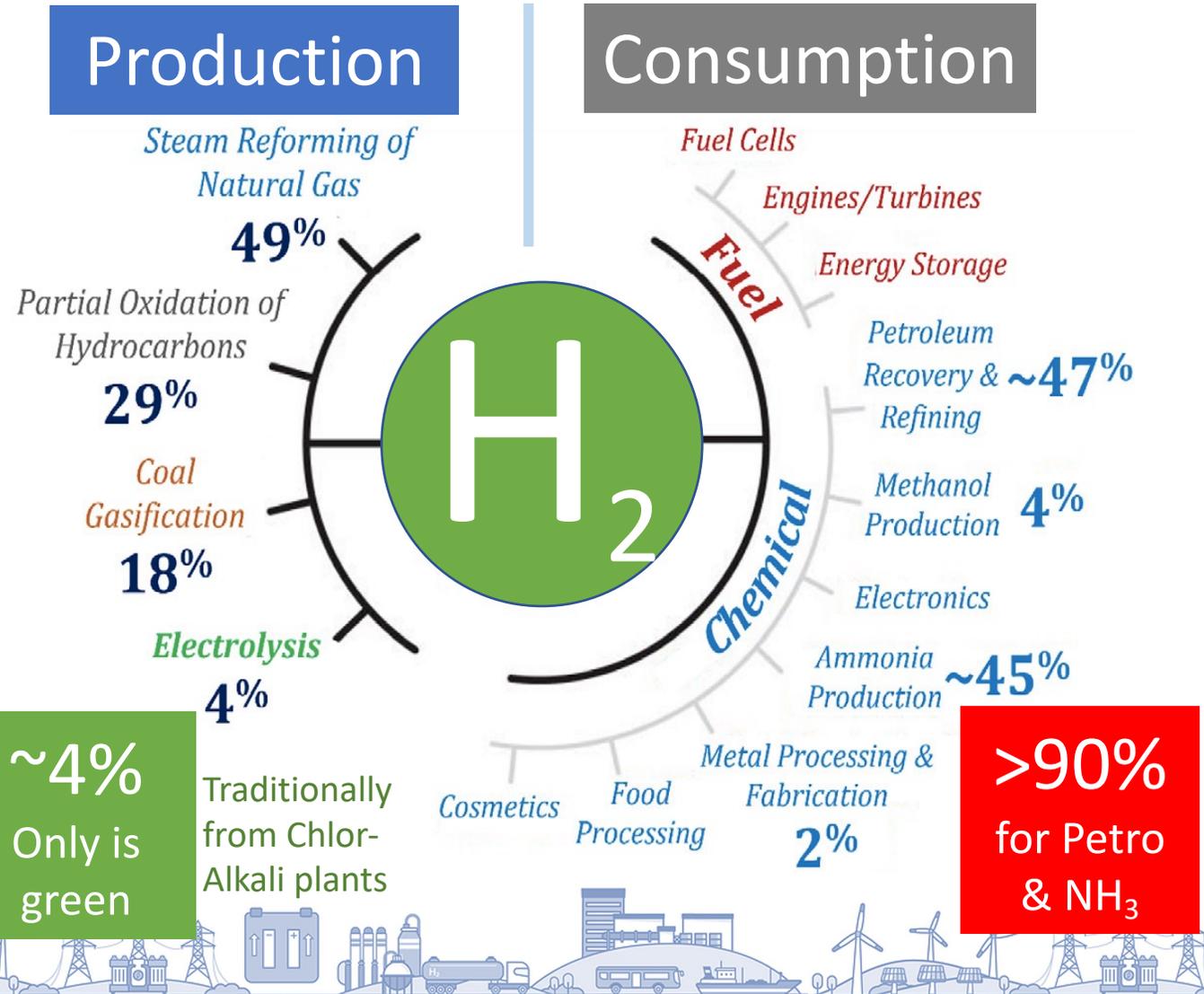


- Hydrogen generation pathways
- Electrolysis & Green H₂ generation
- Electrolyser Technologies: state of the art
- AEM Electrolyser technology: a case study
- Challenges and Opportunities
 - @ core level
 - @ system level
- Testing & Infrastructures
- Summary and Future

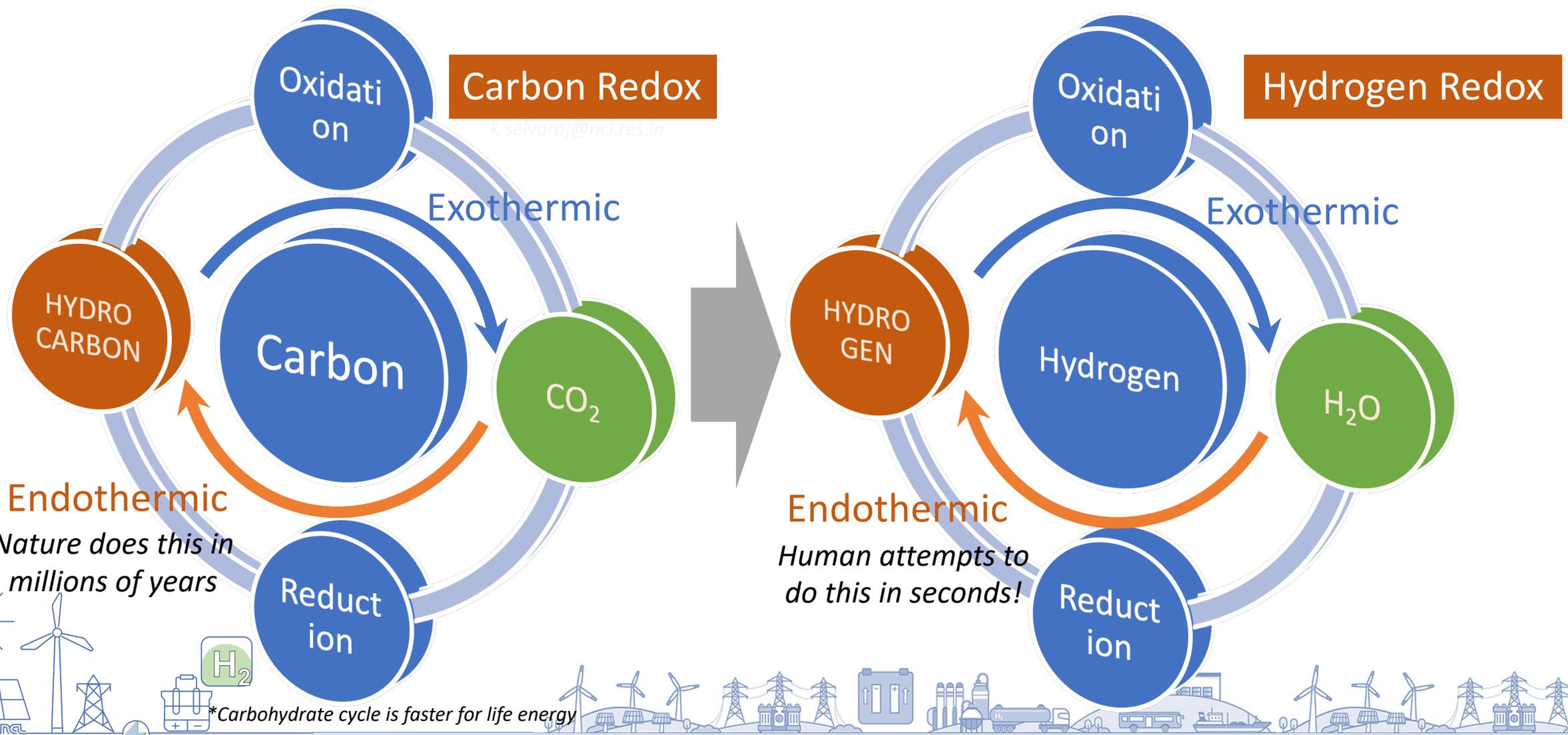


'Implementable Hydrogen Economy': The verticals

- Phase-out fossil fuels & Mitigate climate change
- Low-carbon economy
- Use hydrogen to decarbonize the **hard-to-electrify** sectors. eg., Steel, Cement, Transport
- Develop low cost H₂ technologies
(**Electrolysers**, Storage, Fuel cells etc.)
- Less polluting H₂ generation: CH₄ pyrolysis or SMR with CCS
- Push earlier energy transition



Energy Transition: Simplified



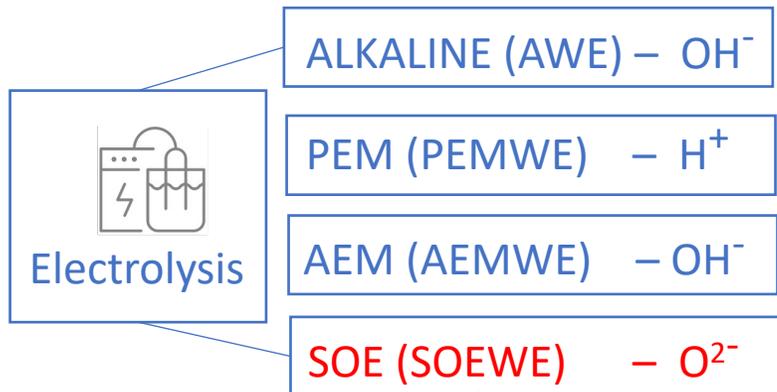
Hydrogen: Sustainable Generation and Projections



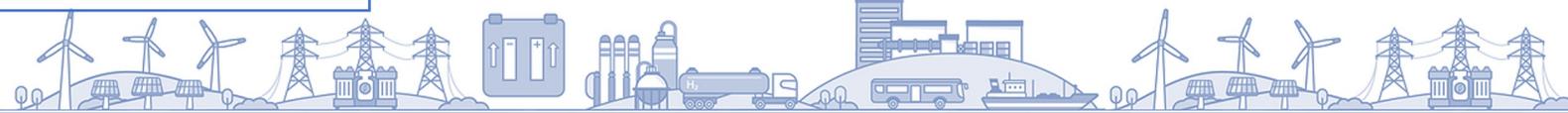
H ₂	Generation	Source	Products	Cost \$/kg	Emission CO ₂
Brown	Gasification	Brown Coal (Lignite)	H ₂ +CO ₂	1.2 – 2.1	High
Black	Gasification	Black Coal (Bituminous)	H ₂ +CO ₂	1.2 – 2.1	High
Grey	Reforming	Natural Gas	H ₂ +CO ₂ released	1.0 – 2.1	Med
Blue	Reform + CC	Natural Gas	H ₂ +CO ₂ 85-95% captured	1.5 – 2.9	Low
Green	Electrolysis	Water	H ₂ +O ₂	3.5 – 5.8	Minimal

Level-play future

	2020				2050			
	Alkaline	PEM	AEM	SOEC	Alkaline	PEM	AEM	SOEC
Cell pressure [bara]	< 30	< 70	< 35	< 10	> 70	> 70	> 70	> 20
Efficiency (system) [kWh/KgH ₂]	50-78	50-83	57-69	45-55	< 45	< 45	< 45	< 40
Lifetime [thousand hours]	60	50-80	> 5	< 20	100	100-120	100	80
Capital costs estimate for large stacks (stack-only, > 1 MW) [USD/kW _{el}]	270	400	-	> 2 000	< 100	< 100	< 100	< 200
Capital cost range estimate for the entire system, >10 MW [USD/kW _{el}]	500-1000	700-1400	-	-	< 200	< 200	< 200	< 300



WE technologies

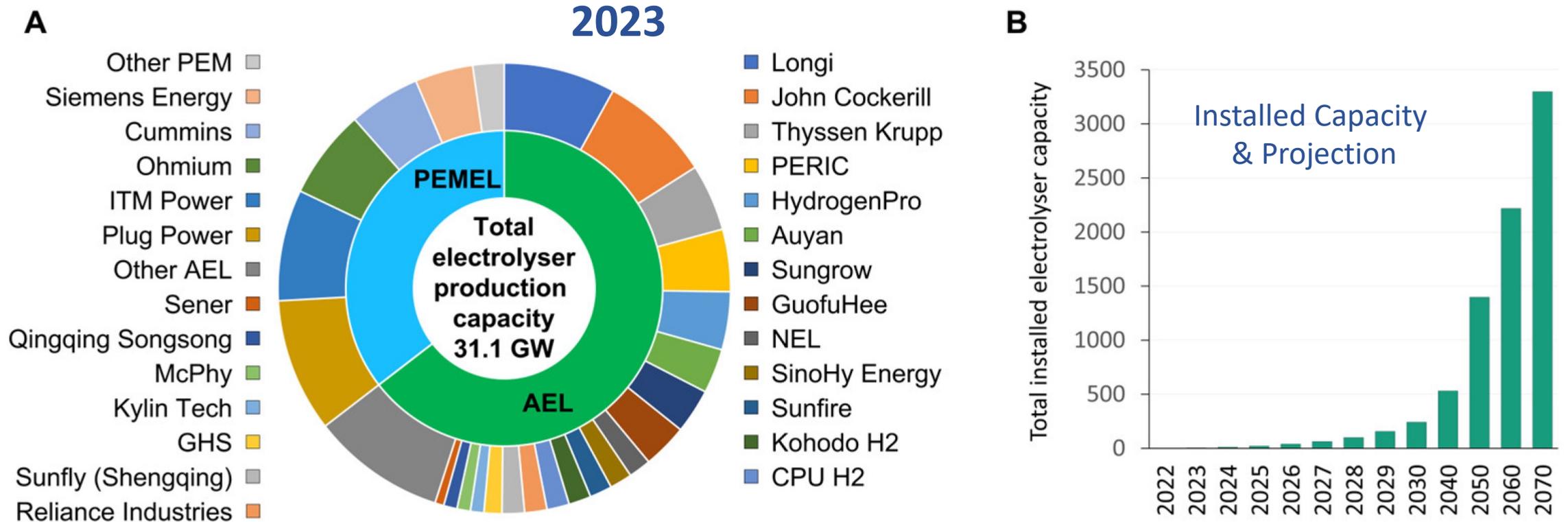


IRENA 2020

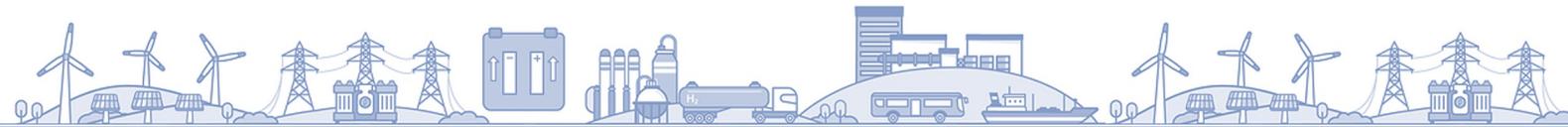
Hydrogen Production: Typical Global Scene



Current electrolyser production capacity by type and manufacturers



Energies 2023, 16(8), 3604



Electrolyser Technologies: State of the art

LT Electrolysis	Alkaline	PEM	AEM
Electrolyte	Aq. KOH (40%)	PEM (Nafion) + (Acidic)	AEM + dil.KOH (<6%)
Cathode	Ni, Ni-Mo alloys	Pt,Pt-Pd	Non-precious metals
Anode	Ni, Ni-Co alloys	RuO2, IrO2	Non-precious metals
Separator	Diaphragm (ZP 500µm)	Nafion 117 (<100µm)	AEM (<50µm)
Cell voltage	1.8 - 2.4 V	1.8 - 2.2 V	1.8 - 2.2 V
Current density	0.2 – 0.4 A/cm2	0.6 – 2.0 A/cm2	0.2 - 1.2 A/cm2
Gas purity (vol%)	>99.5	>99.999	>99.99
Pressure (bar)	1-30	30-75	1-40

CSIR's planned activity
 Collaborating with Industrial partners

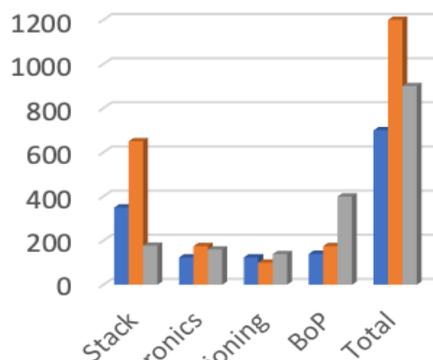


State-of-the-art and future KPIs for all electrolyser technologies.

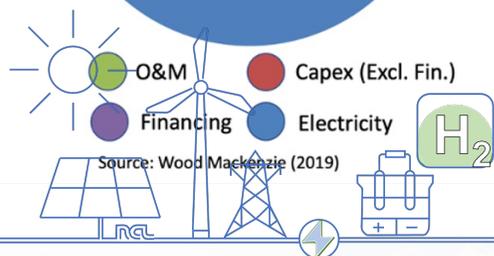
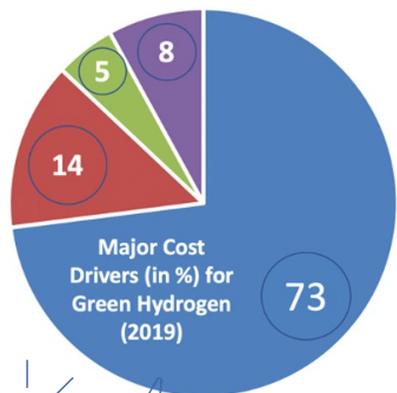
	2020	Target 2050	R&D focus
AEM electrolyzers			
Nominal current density	0.2-2 A/cm ²	> 2 A/cm ²	Membrane, reversion catalysts
Voltage range (limits)	1.4-2.0 V	< 2 V	Catalyst
Operating temperature	40-60°C	80°C	Effect on durability
Cell pressure	< 35 bar	> 70 bar	Membrane
Load range	5%-100%	5%-200%	Membrane
H ₂ purity	99.9%-99.999%	> 99.9999%	Membrane
Voltage efficiency (LHV)	52%-67%	> 75%	Catalysts
Electrical efficiency (stack)	51.5-66 kWh/Kg H ₂	< 42 kWh/Kg H ₂	Catalysts/membrane
Electrical efficiency (system)	57-69 kWh/Kg H ₂	< 45 kWh/Kg H ₂	Balance of plant
Lifetime (stack)	> 5 000 hours	100 000 hours	Membrane, electrodes
Stack unit size	2.5 kW	2 MW	MEA
Electrode area	< 300 cm ²	1 000 cm ²	MEA
Cold start (to nominal load)	< 20 minutes	< 5 minutes	Insulation (design)
Capital costs (stack) minimum 1 MW	Unknown	< USD 100/kW	MEA
Capital costs (system) minimum 10 MW	Unknown	< USD 200/kW	Rectifier

Main cost drivers : green H₂

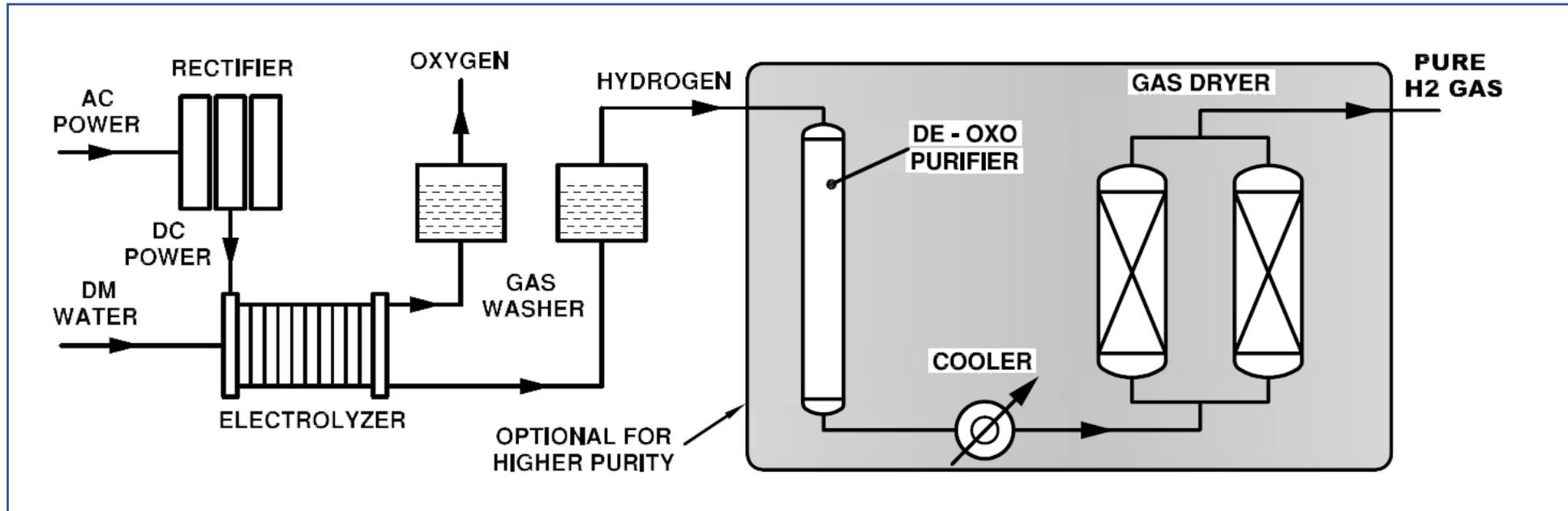
Rough Cost USD @ 1MW
IRENA 2020b



■ Alkaline ■ PEM ■ AEM



Cost: System Technology – Integration & Engineering



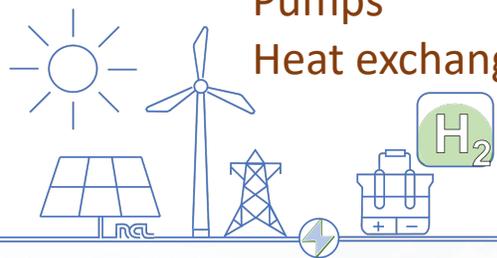
AC-DC converters
Efficient rectifiers
Pumps
Heat exchangers

Gas leak detectors
Dryers
HP components
Accelerated tests

Efficiency influenced by component integration

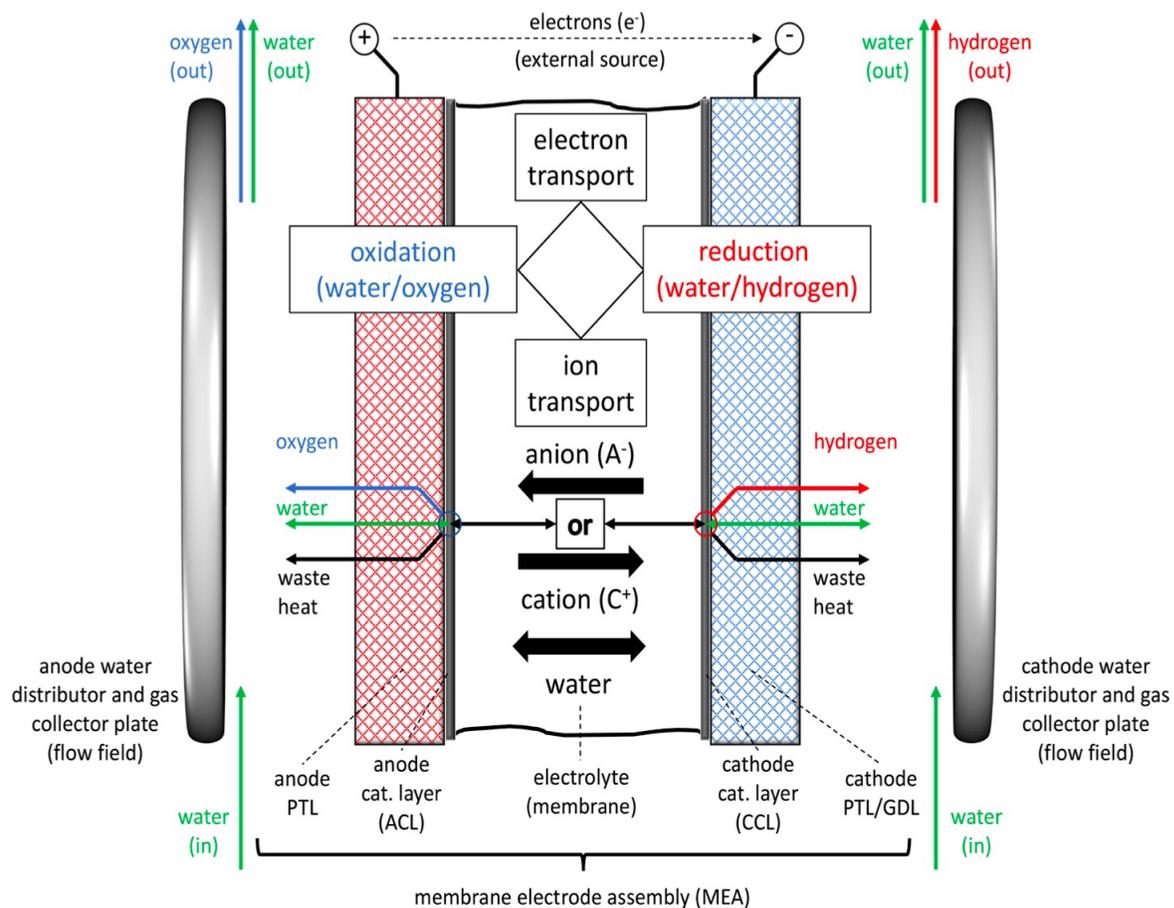
Durability & Life are influenced by affordable MoC & op. conditions

India is fairly comfortable in BoP, Electronics etc.

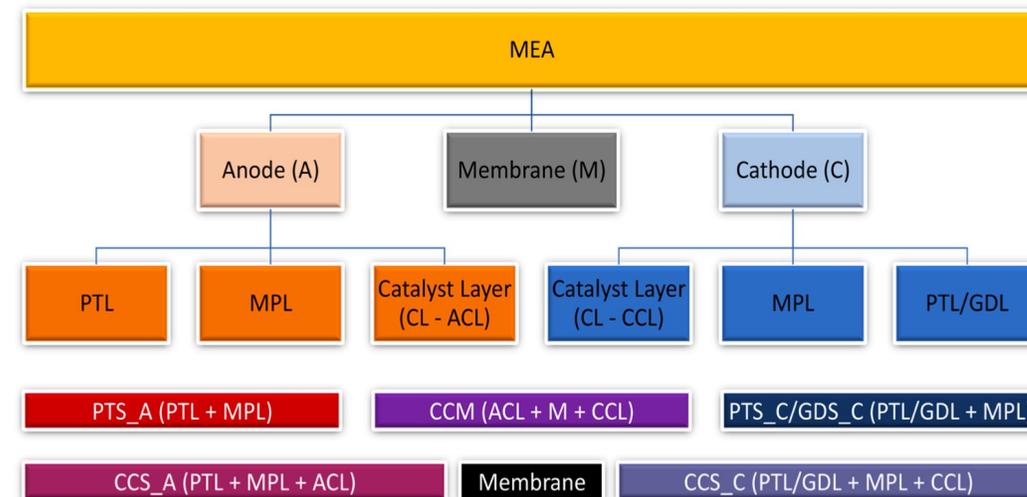


Cost: Core technology: R&D - cheaper & efficient component

Cell Structure



Core Technology



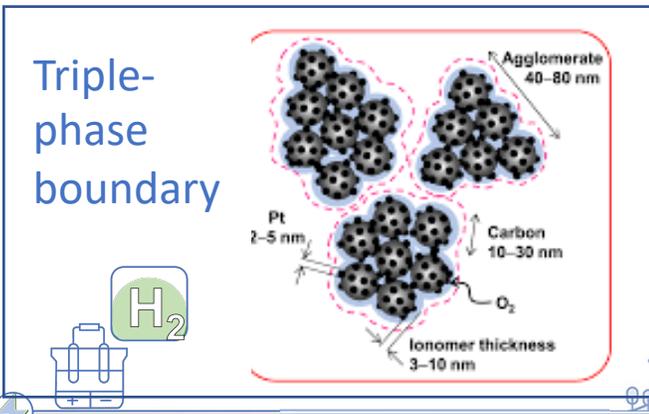
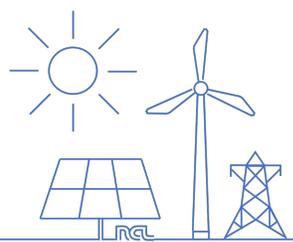
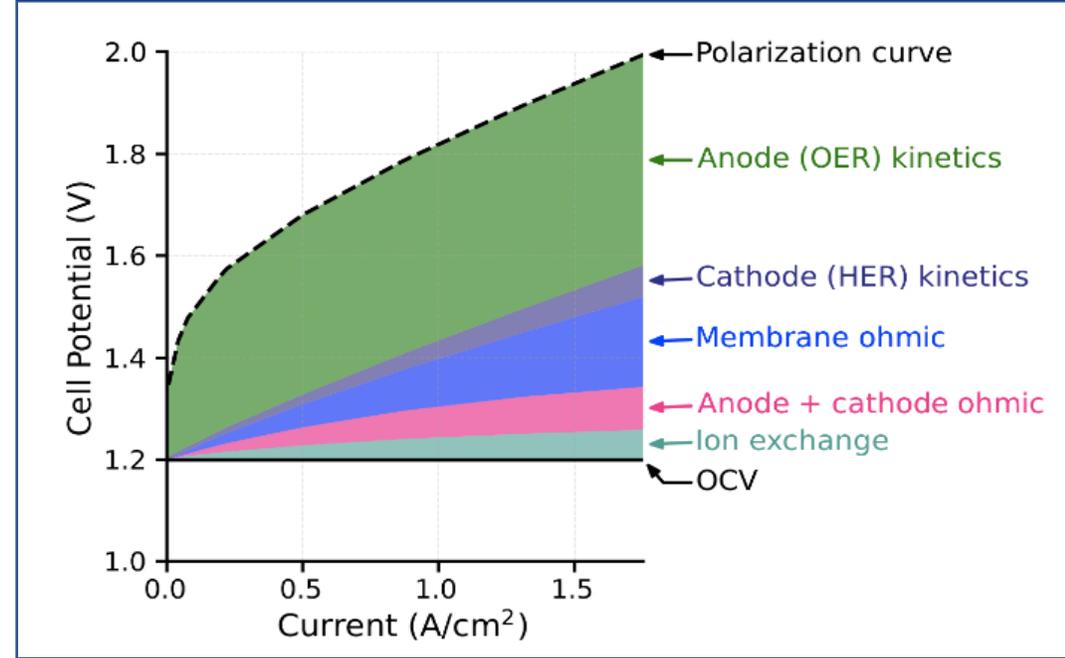
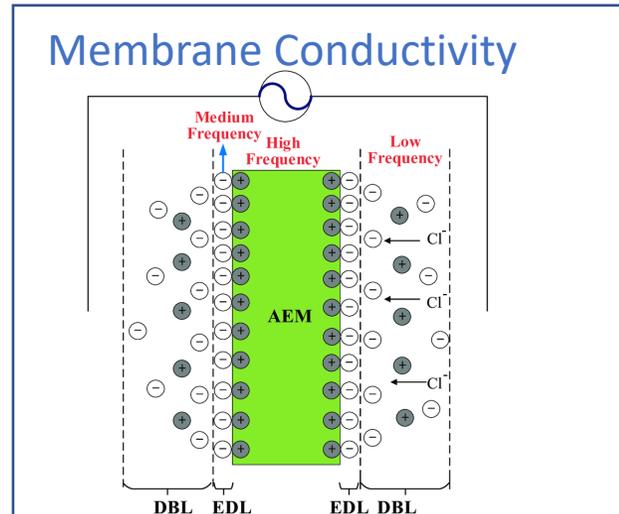
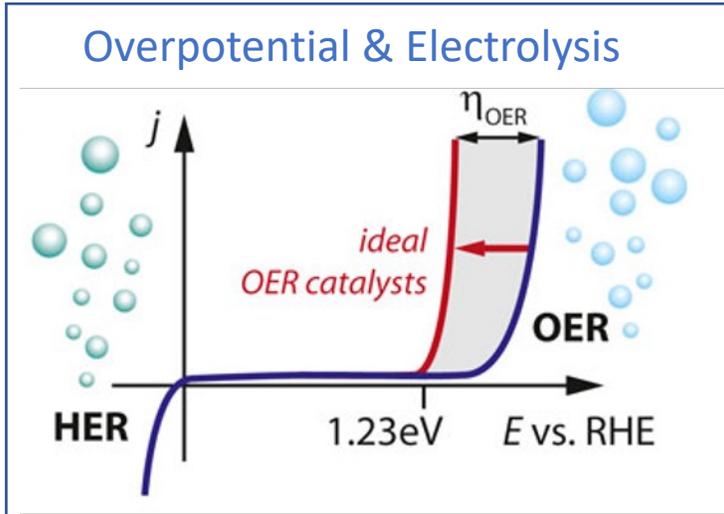
PTL, porous transport layer; MPL, microporous transport layer; GDL, gas diffusion layer; PTS, porous transport system; GDS, gas diffusion system; CCS, catalyst coated substrate; CCM, catalyst coated membrane.

Indigenization of core technology is critical for an Atmanirbhar Bharat

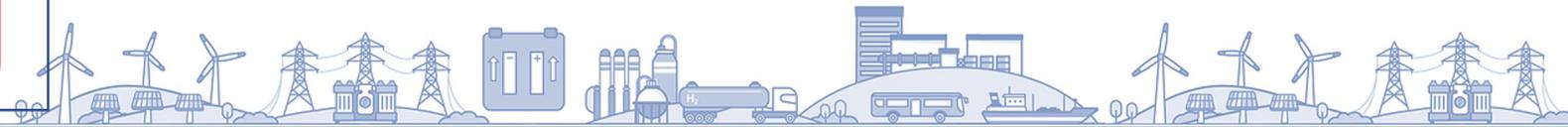
JACS Au 2021, 1, 527-535



Cost: Stack Technology - Thermodynamics & Kinetics



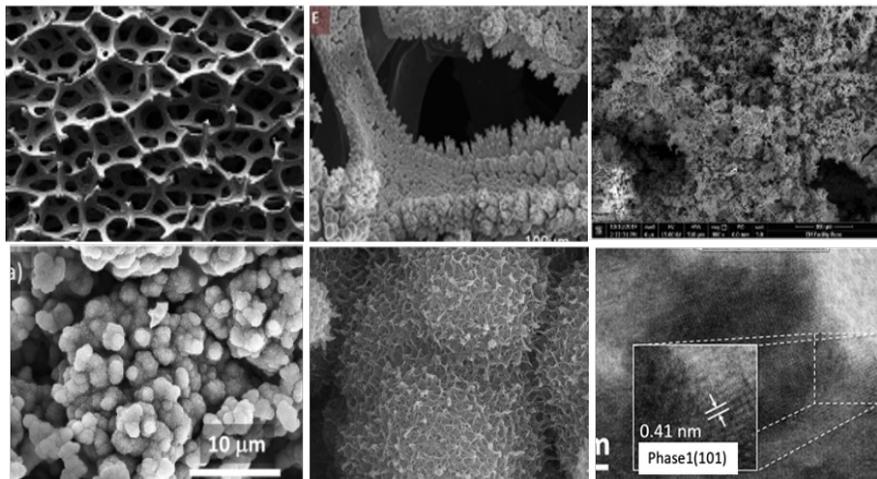
Efficiency influenced by components & their electrochemistry
Durability & Life are influenced by components & their stability
Immense scope for R&D, Game-changing potential



Cost: Stack Technology – Immense R&D opportunity

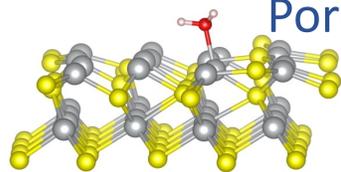


High performing durable non-PGM Catalysts

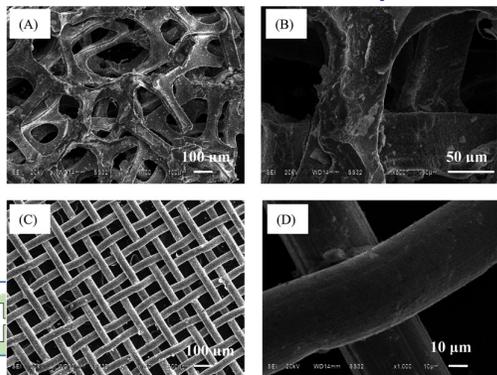


Multilevel hierarchically porous & rough catalysts

Porous metal electrodes/GDLs



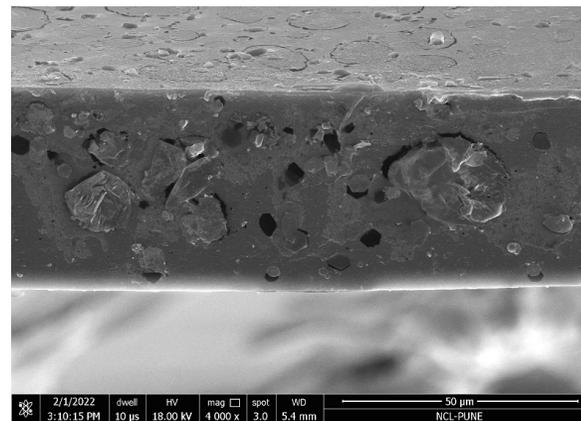
Computational catalyst design



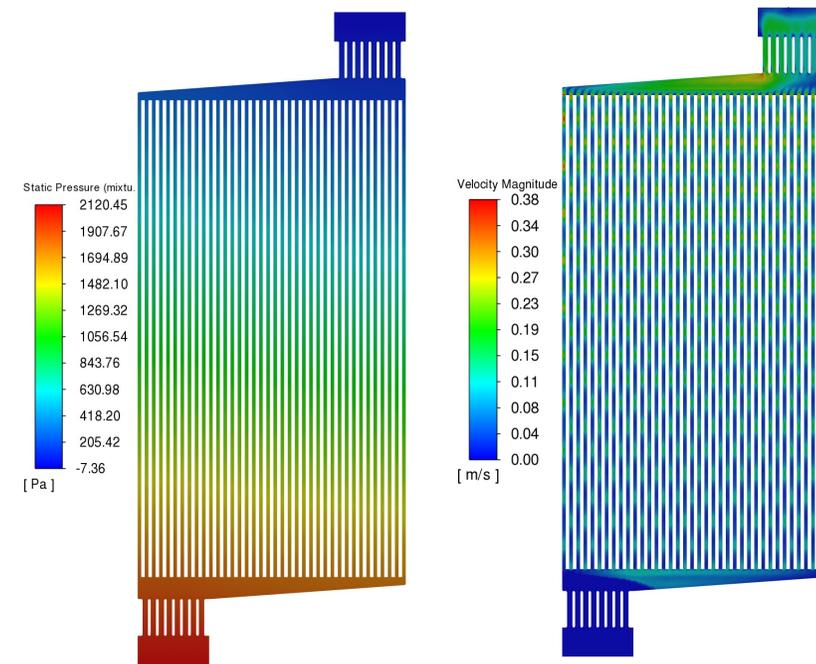
Mass production of membranes



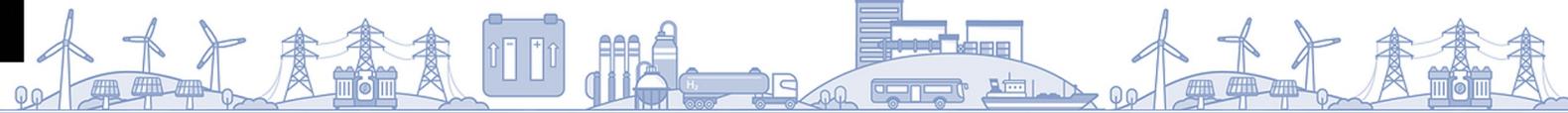
High IEC AE Membranes



CFD simulation based improved flow fields

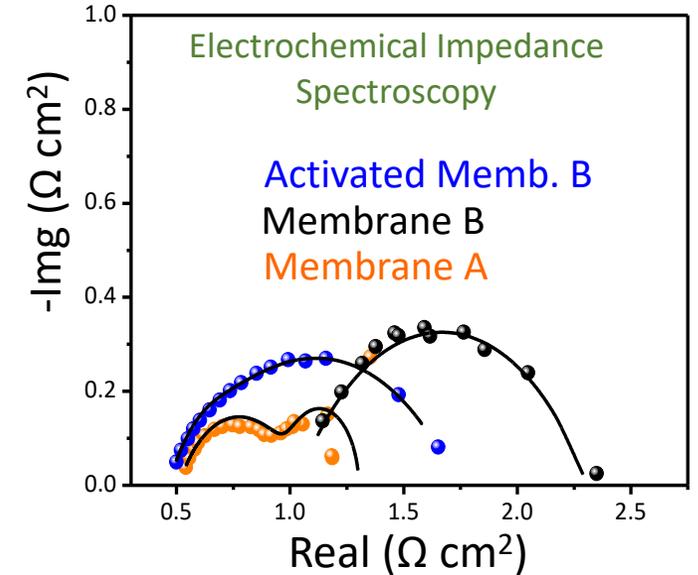
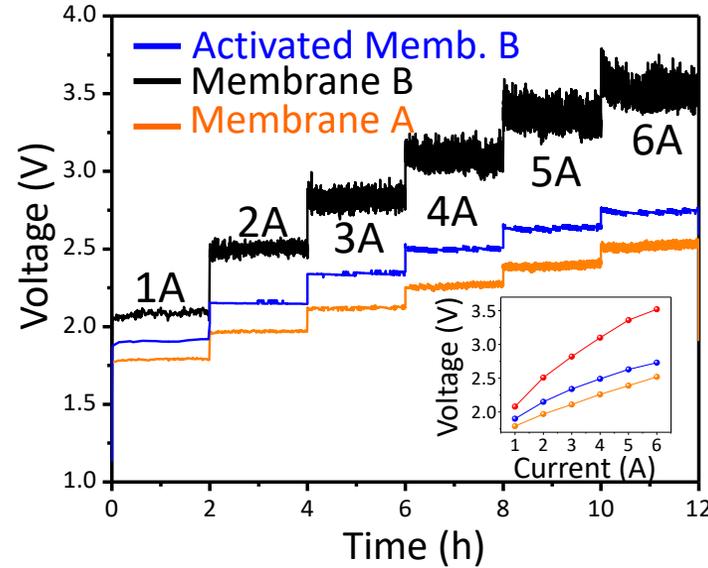
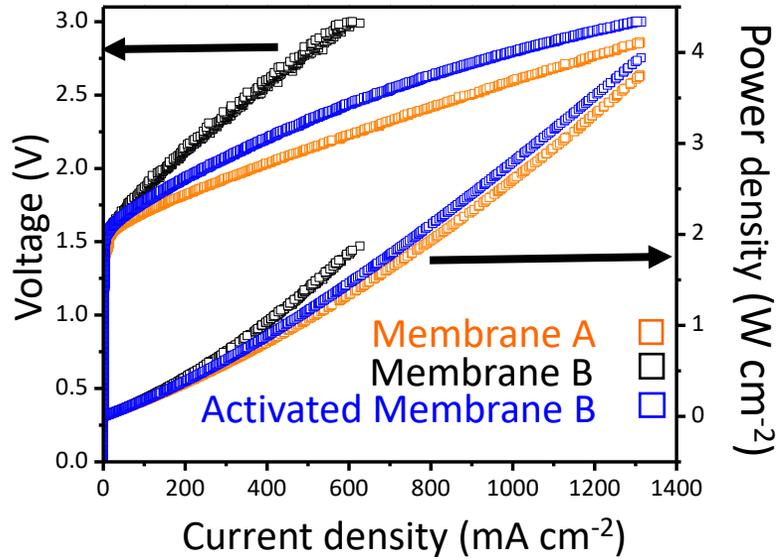


Improved fluid dynamics for better H₂ production



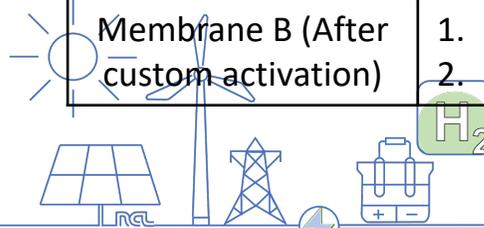
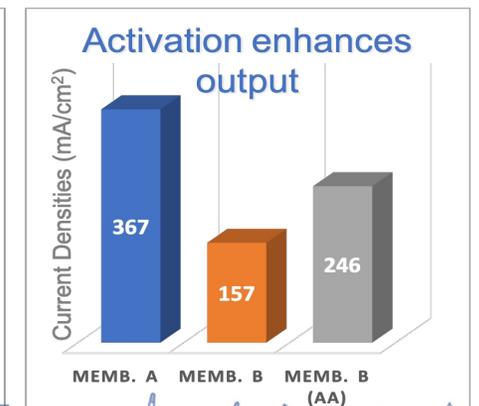
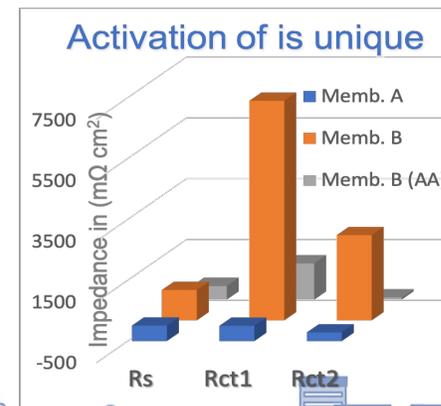
Membrane: Critical component

Catalysts: Ru/C v/s Pt/C @ Room temperature

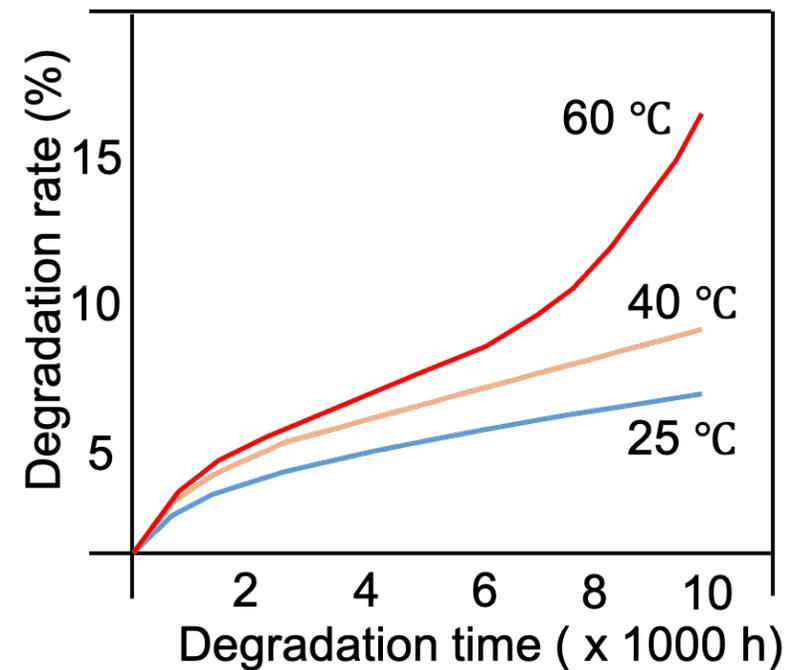
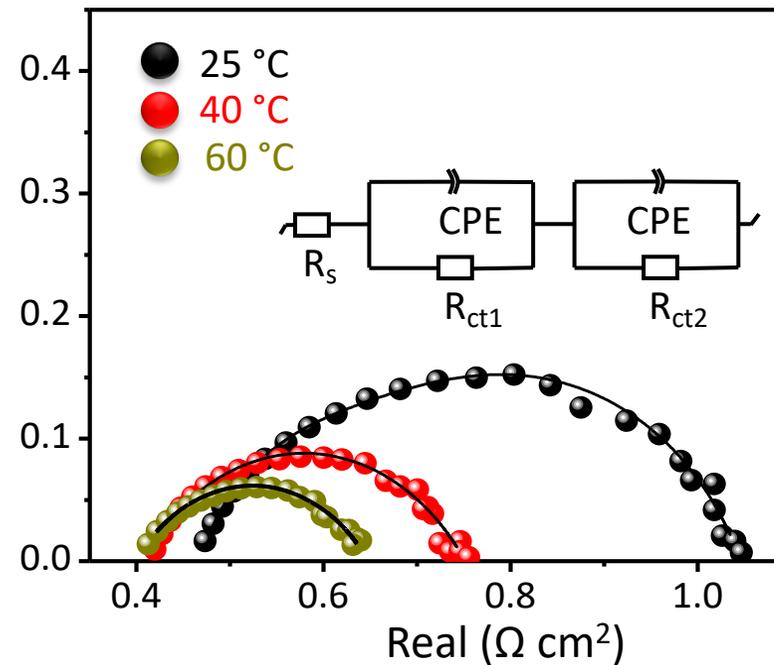
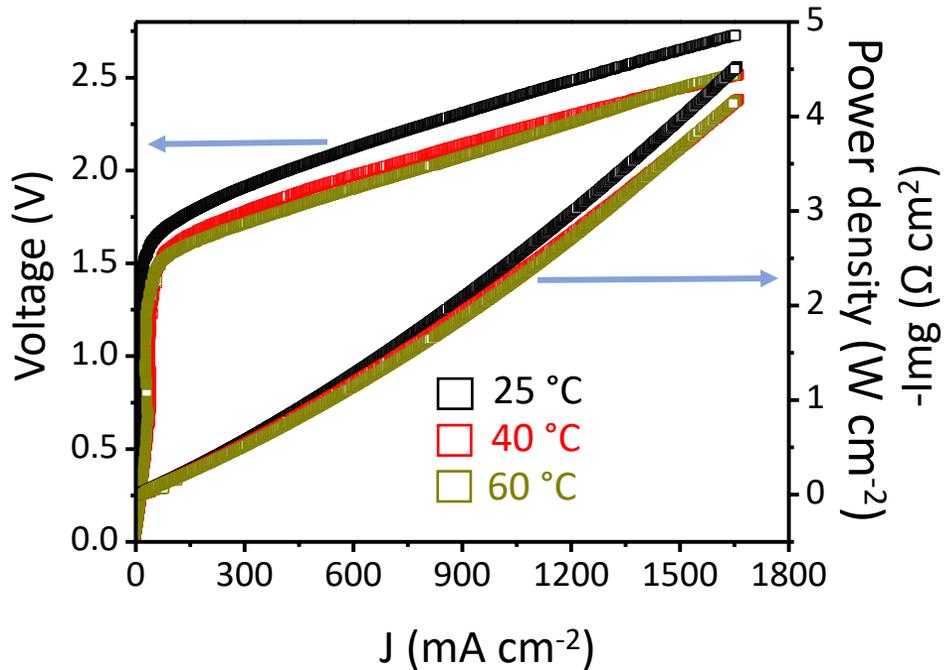


AEM	Typical activation protocol
Membrane A	<ol style="list-style-type: none"> 24h in 1 M KOH (Cl^- ions to OH^- ions) 5h heat treatment @ 55 °C
Membrane B	<ol style="list-style-type: none"> 24h in 1 M KCl (I^- ions to Cl^- ions) 24h in 1 M KOH (Cl^- ions to OH^-) (manufacturer's recipe)
Membrane B (After custom activation)	<ol style="list-style-type: none"> 30h in 1 M KCl (I^- ions to Cl^- ions) 48h in 1 M KOH (Cl^- ions to OH^-) (CSIR NCL's recipe)

Membrane A & B are commercial membranes for AEM

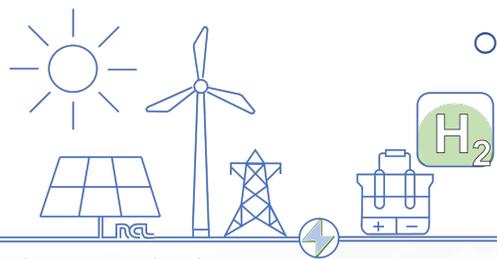


Operating conditions: Performance vs Life - Trade-off



Example case in AEM

- Higher operating temperature enhances the IEC of the membrane and reduces the overpotentials
- Higher potential operation increases the hydrogen production capacity
- Higher operating temperature reduces the life of components and the electrolyser

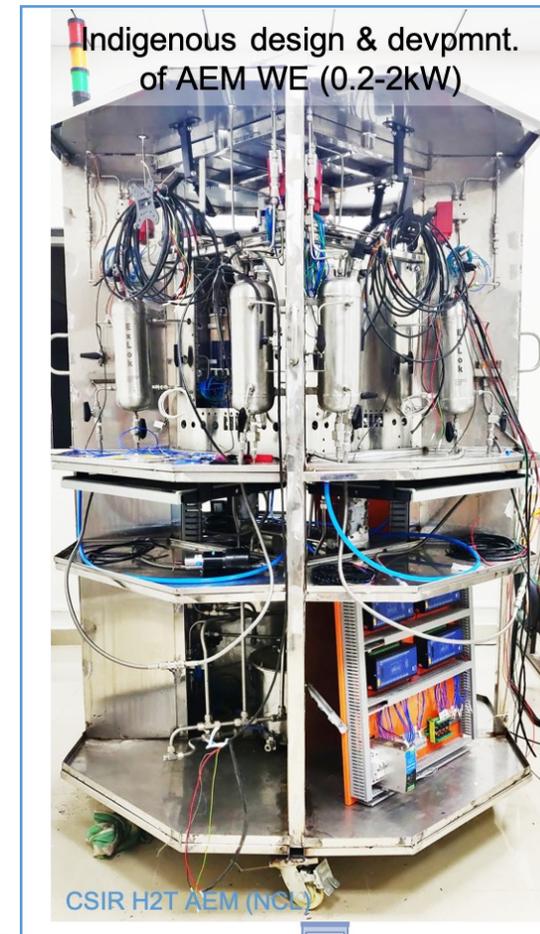


Challenges in testing R&D: Hydrogen Generation Tech.

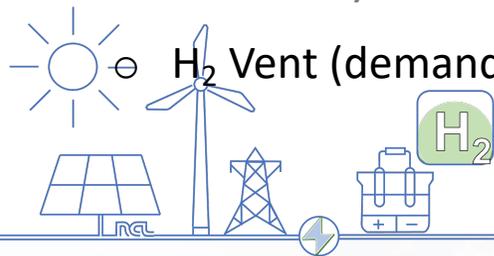
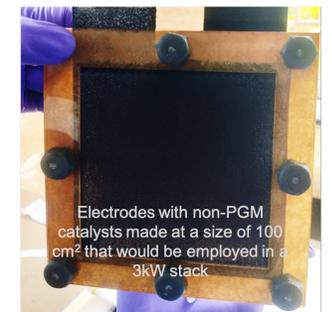
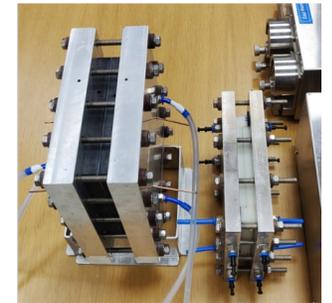


- A handful of international test station manufacturers
- Surging orders, long lead times, long queues
- India: No comm. electrolyser test station manufacturing
- TRL variations (PEM/AEM/SOE): unpredictable test-bench market
- R&D labs are forced to fabricate own test stations (extra load)
- Import dependency: components (EIS) & accessories
- Poor supply chain: components & consumables
- Expensive power supplies: AC-DC or DC-DC converters
- Affordable & reliable H₂ leak detection devices
- Lab Safety: Continuous & larger quantities of H₂ & O₂ generation
- H₂ Vent (demands flash arrestors / N₂ for dilution)

CSIR's first indigenous AEM Water Electrolyser



100 cm²
square cell



Challenges in testing manufacturing: Hydrogen Generation



- Testing hydrogen production at MW level needs new facilities
- Such facilities/protocols are unprecedented
- Key aspects: Electricity, Water, Gas analysis (H₂ & O₂)
- Large-scale utilities: Power supply (@MW scale)
- Low voltage (400V)-high current(2.5kA): needs new excl. safety
- Safety: Large quantities of H₂ & O₂ (H₂ sink)
- Large-scale gas flow: Limiting the range of MFM
- Testing cost: Affordable Electricity & Remunerated H₂ sink
- H₂ sink1: Cold Vent (demands 24X of N₂ for dilution)
- Venting large H₂ is eventually not feasible (GWP of H₂ is 5.8)
- H₂ sink2: Flare (to be alive, demands CNG: emissions & soot!)
- H₂ sink3: Captive consumption FC (high CAPEX, H₂ purity!)
- H₂ sink4: Chemical industry via pipeline (CAPEX, purity!)



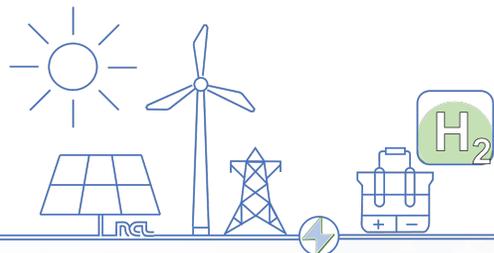
Electrolyser test field at the Hydrogen Lab Leuna. (1) Concrete slab for placing devices under testing (DUT), (2) safety cold vent, (3) connection to Linde H₂ pipeline as H₂ sink, (4) medium-voltage power supply, (5) programmable logic controller and control room interface, (6) individual utility interface (IUI). Insert: Close-up of the IUI. (7) H₂ and O₂ analysis, (8) N₂ supply, (9) compressed air supply, (10) H₂ product output, (11) process water supply, (12) deionized water supply, (13) low-pressure steam supply. © Fraunhofer IWES.

- Location of test facility: Non-residential (Explosion Safety)
- TRL-based test variations: 'One-size-does-not fit all'
- AWE, PEM, SOEL, PC, PEC, AEM

Summary & Future



- Affordable electrolyser technology development is key to realizing Energy Transition
- Apart from affordable renewables, R&D at core technology is critical to reducing LCOH of H₂
- India should focus on building R&D infrastructure & testing capabilities
- Component development and manufacturing are keys for India to be self-reliant in H₂ generation & export
- AEM seems to be a game changer in electrolyser technology; Focused activity will help to realize it.
- Setting safety & standards will boost the tech. development and market penetration
- Incentivizing PPP mode tech development will help to fill up the 'Valley of Death' in mid-range TRLs
- The global electrolyser industry is set to grow exponentially, and India should not miss the bus.



INTERNATIONAL CONFERENCE ON GREEN HYDROGEN 2023

THANK YOU

For kind attention



MINISTRY OF NEW
AND RENEWABLE ENERGY
GOVERNMENT OF INDIA



MINISTRY OF PETROLEUM
AND NATURAL GAS
GOVERNMENT OF INDIA

OFFICE OF THE PRINCIPAL
SCIENTIFIC ADVISER TO THE
GOVERNMENT OF INDIA



CSIR National Chemical Laboratory

Laboratory for Advanced Science and Enduring Technologies

Catalysis & Inorganic Chemistry Division

Pune 411008, India <https://academic.ncl.res.in/k.selvaraj/>

